

Cartoons of George Eastman

## Influence of the Nature of Developers on the Size of Grains of Reduced Silver in Photographic Plates

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It has been admitted after various studies on the subject that the grains of reduced silver obtained by means of different developers acting on photographic plates are always of uniform size notwithstanding the nature of the reducing substance used. However, Abney has found that the grains of silver in an over-exposed plate are finer than those of the same plate normally exposed, also that the addition of great quantities of alkaline bromides seems to increase the size of the grains.

Having obtained with certain substances of weak developing power very transparent images of a color notably different from that of usual silver images, we have thought that there might be a relation between the size of the grains of reduced silver and the color of the image.

To ascertain the correctness of this hypothesis, we have compared the size of the grains obtained by the most usual developers prepared according to the normal formula, and we have also studied on the same developers the influence of dilution, of duration of development, temperature and alkalinity. Finally, we have examined the modifications determined by variation in the length of exposure, as well as the results obtained by under or over development.

### AN INFLUENCE OF DEVELOPING SUBSTANCES

We have prepared developers with the following reducing substances:

Hydrochinon, pyrogallic acid, amidol, metol, metochinon, eikonogen, rodinal, glycin, edinol, adarol, ortol, hydramin, pyrocatechin.

Outside of the preceding compounds, which comprise about all the developing substances practically used, we have experimented with other substances which, on account of their weak developing power, have not yet been generally utilized. They were particularly interesting to us, for they produce a silver image of a special color. These substances, the paraphenylenediamin and the orthoamidophenol, are used in presence of sodium sulphite alone. They determine the formation of a silver image of great transparency, having a brownish tint when the rays of light are admitted through them, while they look grey by reflected light, their aspect being quite similar to that of images obtained in the development of collodion plates.

The composition of the different developers we have used is as follows:

<i>Hydrochinon Developers</i>	
Hydrochinon .....	10
Sodium sulphite anhydrous .....	40
Sodium carbonate anhydrous .....	55
Water .....	1000
Or	
(a) Hydrochinon .....	10
Anhydrous sodium sulphite .....	30
Water .....	500
(b) Tribasic sodium phosphate .....	80
Warm water .....	500
Mix 50 of A and 55 of B.	
<i>Pyrogallic Acid Developer</i>	
(a) Water .....	500
Anhydrous sodium sulphite .....	50
Pyrogallic acid .....	14
(b) Water .....	500
Sodium carbonate .....	50
Water .....	1000
Sulphite .....	100
Sodium phosphate .....	90
Pyrogallic acid .....	40
Or	
(a) Water .....	1000
Sulphite .....	100
Pyrogallic acid .....	40
(b) Acetone .....	—
Mix water, 75; (a) 25; acetone, 10	
<i>Eikonogen Developer</i>	
Anhydrous sodium sulphite .....	30
Potassium carbonate .....	30
Eikonogen .....	35
Water .....	1000

<i>Metol-hydrochinon Developer</i>	
(a) Distilled water .....	400
Metol .....	2
Anhydrous sulphite .....	20
Hydrochinon .....	35
(b) Distilled water .....	400
Potassium carbonate .....	40
Mix equal parts of A and B.	
<i>Metol Developer</i>	
(a) Water .....	1000
Anhydrous sulphite .....	75
Metol .....	10
(b) Water .....	1000
Anhydrous sodium carbonate .....	75
Mix 50 of A and 25 of B.	
<i>Rodinal Developer</i>	
Water .....	1000
Anhydrous sulphite .....	75
Caustic lithia .....	5
Paramidophenol (free base) .....	10
<i>Hydramin Developer</i>	
Water .....	1000
Hydramin .....	5
Anhydrous sulphite .....	15
Caustic lithia .....	3
<i>Pyrocatechin Developer</i>	
(a) Water .....	300
Anhydrous sulphite .....	20
Pyrocatechin .....	10
(b) Water .....	500
Potassium carbonate .....	10
<i>Adarol Developer</i>	
Anhydrous sulphite .....	100
Potassium carbonate .....	150
Water .....	500
Adarol .....	25
<i>Edinol Developer</i>	
Edinol .....	5
Anhydrous sulphite .....	20
Sodium phosphate .....	30
Water .....	500
<i>Glycin Developer</i>	
(a) Water .....	1000
Anhydrous sulphite .....	15
Glycin .....	10
(b) Water .....	500
Potassium carbonate .....	100
Mix 100 of A and 25 of B.	
<i>Amidol Developer</i>	
Water .....	1000
Amidol .....	5
Anhydrous sulphite .....	30
<i>Metochinon Developer</i>	
Water .....	1000
Sodium sulphite .....	60
Metochinon .....	9
Water .....	1000
Anhydrous sulphite .....	60
Aceton .....	30
Or	
Metochinon .....	9
Water .....	1000
Anhydrous sulphite .....	60
Metochinon .....	9
Caustic lithia .....	6
<i>Paraphenylenediamide Developer</i>	
Water .....	1000
Paraphenylenediamin .....	10
Anhydrous sulphite .....	60
<i>Orthoamidophenol Developer</i>	
Water .....	1000
Amidophenol .....	10
Anhydrous sulphite .....	60

With all these different solutions, we have developed Lumiere plates, blue label, covered with the same emulsion and consequently were sure of starting with silver bromide having grains of exactly the same size. All the plates exposed under the same conditions, were developed in the different solutions kept at the same temperature, 20° C. and the development

was carried on in all cases, so as to obtain images of a practically equal intensity.

After washing these plates completely, we applied on a small portion of the film a little hot water, so as to dissolve the gelatin, choosing for this treatment an opaque part of the image, in order to get a great quantity of reduced silver. The film of a plate developed with pyrogallic acid becoming insoluble, it was in that case necessary to break up the gelatin by heating with a solution of caustic alkali.

With the gelatinous solution, well stirred, and containing in suspension the reduced silver, microscopic slides were prepared. The whole series of negatives corresponding to the various developers were treated in the same way, and the microscopic preparations were photographed under exactly the same conditions of enlargement. From the examination of the prints, the following results appeared:

1° The size of the grains of silver reduced by divers developers ordinarily used was practically the same with all the reducing agents.

2° Paraphenylenediamin and orthoamidophenol developers produce very transparent and characteristically colored silver images with grains much finer than those produced by other developing substances.

### B. INFLUENCE OF DILUTION OF DEVELOPER AND DURATION OF DEVELOPMENT.

To study the influence of the dilution of developer, we have employed a normal developing solution of hydrochinon and sodium carbonate, and another one of diaminodophenol (Amidol). In a first experiment the solutions were diluted with an equal volume of water, and in a second trial with ten times their volume of water. In both cases the development was carried on so as to get an image of the same intensity.

We have also developed the plates in a vertical developing dish, using the following very weak solution:

<i>Slow Metochinon Developer</i>	
Water .....	4000
Metochinon .....	5
Anhydrous sulphite .....	50
Potassium bromide, (10 per cent) .....	5

<i>Slow Amidol Developer</i>	
Water .....	2000
Amidol .....	3
Anhydrous sulphite .....	9

Finally we developed two plates in a normal hydrochinon developer, stopping the development for the first one as soon as the image was clearly visible, while the other was left in the bath and developed thoroughly so as to get a very dense image. The same experiment was repeated, using a normal amidol developer.

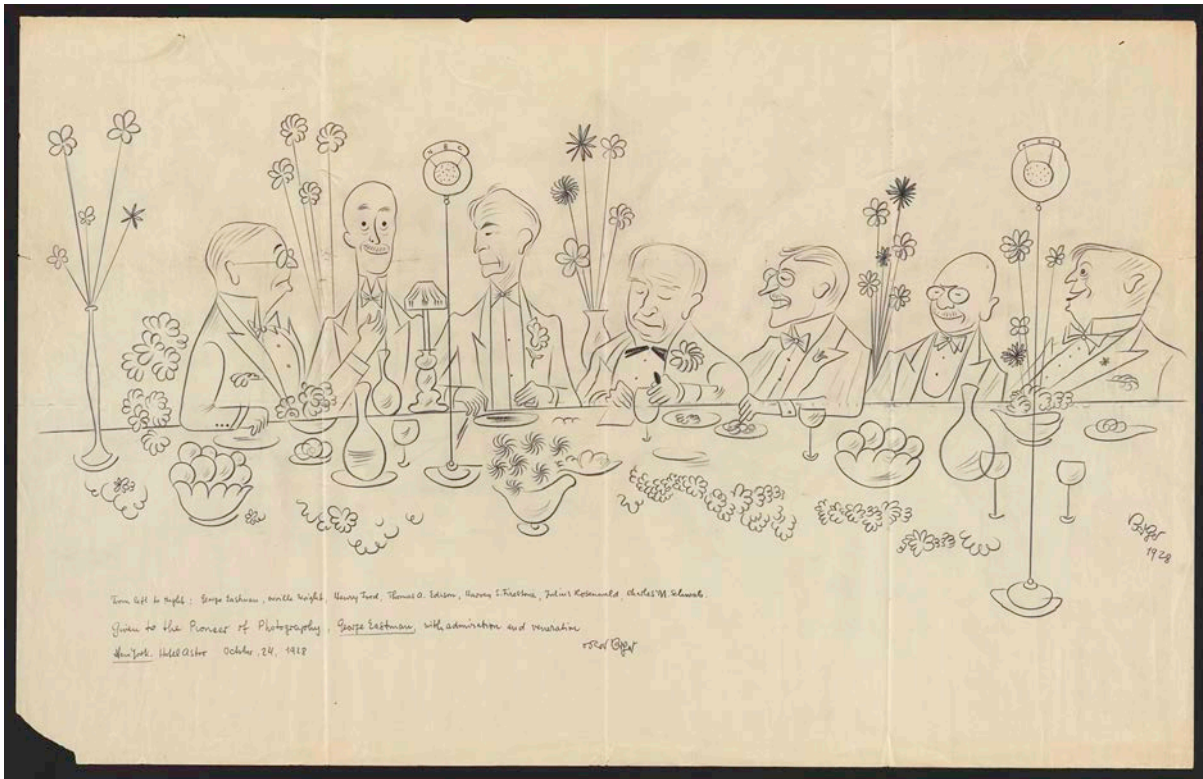
The plates thus obtained were completely washed and each one of them treated with hot water as previously described to prepare a microscopic slide, which was photographed. Examination of the microphotographs thus obtained shows that the size of the grains differs only in case of slow development, when the grains seem to be slightly finer than in the other tests.

### C. INFLUENCE OF TEMPERATURE AND ALKALINITY.

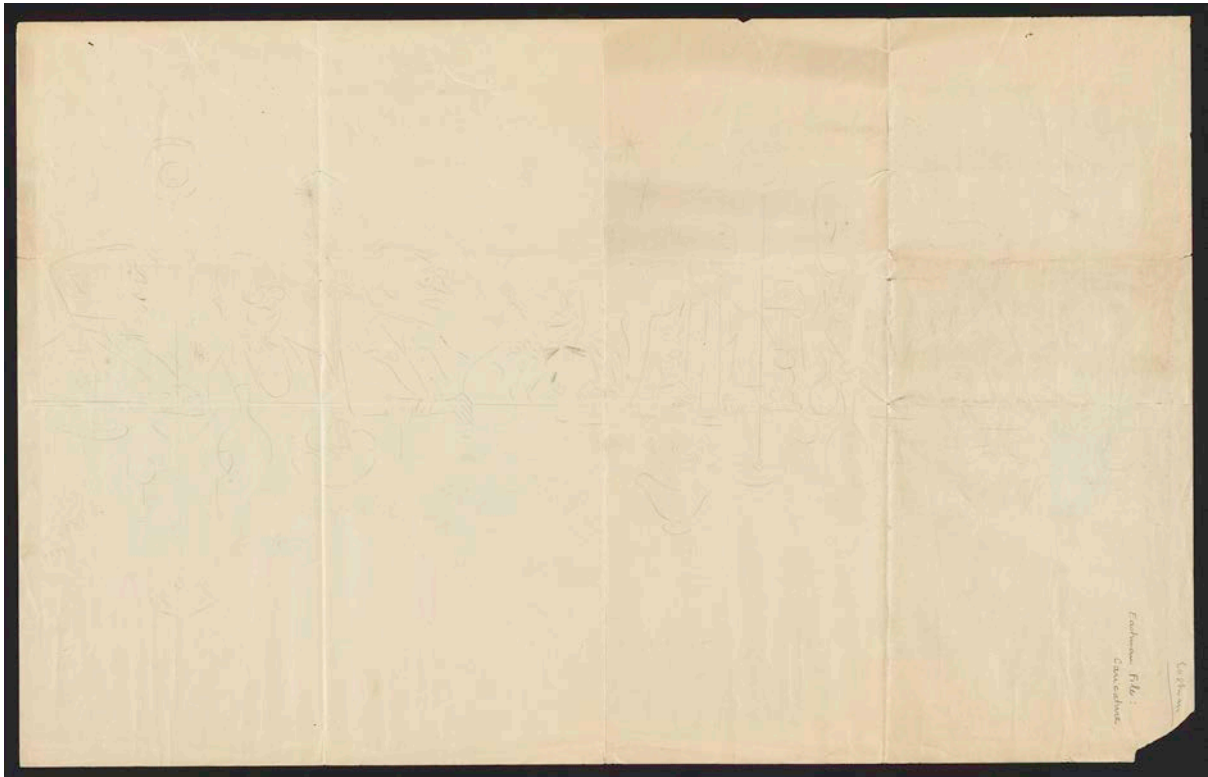
We have made a series of experiments with hydrochinon and amidol developers maintained during the whole development at temperatures of 5, 15, 25 and 35° C., bringing in all cases the images to about the same final density.

We have examined also for the same temperature the influence of alkalinity, increasing or diminishing the quantity of sodium carbonate in a series of experiments, and adding to the normal amidol developer an increasing amount of sodium bisulphite, so as to have an acid developer and thus increase in considerable proportion the normal duration of development.

The plates obtained were treated and examined as previously described, but in no case



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